

PENGARUH SURFAKTAN CTMA-Br (Setiltrimetilamonium-Bromida) PADA
SINTESIS LEMPUNG TERPILAR $\text{SiO}_2/\text{TiO}_2$ SERTA APLIKASINYA SEBAGAI
FOTOKATALIS DEGRADASI *METANIL YELLOW*

Oleh :

Siwi Harning Pambudi
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RINGKASAN

Telah dilakukan penelitian tentang pengaruh penambahan surfaktan CTMA-Br (setiltrimetilamonium-bromida) dengan memvariasi komposisi pada sintesis lempung terpilar silika (SiO_2) dari lempung alam Pontianak serta diuji aktivitas katalitiknya untuk fotokatalis degradasi *metanil yellow*. Variasi komposisi CTMA-Br dilakukan untuk mengetahui hubungan penambahan surfaktan CTMA-Br terhadap *basal spacing*, kristalinitas, luas permukaan spesifik, dan distribusi pori lempung terpilar silika yang dihasilkan. Titanium dioksida (TiO_2) merupakan senyawa semikonduktor yang telah digunakan sebagai katalis dalam proses degradasi senyawa berwarna. Pada penelitian ini lempung terpilar silika yang dihasilkan digunakan sebagai media pengembang untuk mendistribusikan katalis TiO_2 .

Lempung terpilar silika (SiO_2) disintesis dengan melakukan pertukaran kation menggunakan surfaktan CTMA-Br (variasi komposisi) dan interkalasi TEOS (tetraetilortosilikat)-dodesilamin pada daerah antarlapis lempung serta kalsinasi lempung terinterkalasi TEOS pada temperatur 600 °C. Karakterisasi hasil meliputi *basal spacing* dan kristalinitas dengan menggunakan difraksi sinar-X, serta luas permukaan spesifik dan distribusi pori dengan menggunakan *Surface Area Analyzer* (metode BET). Selanjutnya lempung terpilar hasil sintesis diimpregnasi dengan titanium dioksida (TiO_2). Uji aktivitas katalitik lempung terpilar teremban TiO_2 dilakukan terhadap proses fotokatalis degradasi *metanil yellow*. Analisis hasil fotokatalis degradasi dilakukan dengan menggunakan spektrofotometer UV-Vis.

Lempung terpilar silika (SiO_2) yang dihasilkan dengan memvariasi komposisi surfaktan CTMA-Br mempunyai *basal spacing* semakin menurun, kristalinitas semakin meningkat, luas permukaan spesifik semakin menurun dan distribusi pori yang lebih merata dan ukurannya lebih seragam seiring dengan semakin banyak CTMA-Br yang ditambahkan. Aplikasi sebagai fotokatalis degradasi *metanil yellow* menunjukkan lempung terpilar silika (SiO_2) teremban TiO_2 mampu mendegradasi *metanil yellow*. Semakin banyak jumlah TiO_2 yang teremban dalam lempung terpilar silika, semakin besar aktivitas katalitiknya.

SUMMARY

A research on the effect of addition of CTMA-Br (cetyltrimethylammonium-bromide) surfactant on the synthesis of silica pillared clay (SiO_2) from Pontianak natural clay had been done. The research was done by varying the composition of surfactant to find out the influence of CTMA-Br addition on the basal spacing, crystallinity, specific surface area, and the pore distribution of the synthesized pillared clay. The clay was then tested for its catalytic activity in the photocatalysis of methanil yellow degradation. Titanium dioxide (TiO_2) is a semiconductor material which has been widely used as catalyst in coloring agent degradation. In this research, the produced silica pillared clay was used as impregnated media to distribute TiO_2 catalyst.

Silica pillared clay (SiO_2) was synthesized by performing ion-exchange process using CTMA-Br surfactant, TEOS (tetraethylorthosilicate)-dodecylamine intercalation on interlayer area of clay and calcination of TEOS-intercalated clay at 600°C . The characterization was carried out by analyzing the basal spacing and crystallinity of pillared clay by X-Ray Diffraction (XRD) and the specific surface area and pore distribution by surface area analyzer (BET method). The pillared clay was then impregnated by TiO_2 . The catalytic activity of pillared clay impregnated TiO_2 was tested against the photocatalysis of methanil yellow degradation. The photocatalysis result was analyzed by UV-Vis spectrophotometer.

The research data showed that the basal spacing and the specific surface area of clay decreased as the addition of CTMA-Br increased. On the other hand, its crystallinity and pore distribution increased by the increasing of surfactant concentration. The application of pillared clay impregnated TiO_2 showed that it was able to decolorize methanil yellow. The more TiO_2 impregnated on the pillared clay, the greater its catalytic activity.

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