

SINTESIS DAN KARAKTERISASI LEMPUNG TERPILAR TiO₂

Oleh:

Slamet Agus Purnomo

J2C001176

RINGKASAN

Indonesia mempunyai bahan alam berupa tanah lempung yang berlimpah dan belum dimanfaatkan secara optimal. Beberapa penelitian telah dilakukan untuk memanfaatkan lempung menjadi material baru yang lebih berguna misalnya sebagai katalis atau adsorben. Dengan melakukan modifikasi strukturnya, lempung dapat diolah menjadi material baru dengan sifat-sifat fisik dan kimia lebih baik dari sebelumnya.

Sintesis lempung terpillar TiO₂ dilakukan dengan cara interkalasi larutan pemilar titanium pada lempung dilanjutkan dengan kalsinasi. Suhu kalsinasi divariasikan untuk melihat pengaruhnya terhadap *basal spacing*, stabilitas termal, angka keasaman, situs asam Brønsted-Lewis dan luas permukaan lempung terpillar.

Karakterisasi *basal spacing* dan stabilitas termal menggunakan XRD, angka keasaman dan situs asam Brønsted-Lewis menggunakan adsorpsi piridin/IR dan luas permukaan menggunakan metode adsorpsi gas nitrogen melalui persamaan BET. Hasil analisis menunjukkan bahwa lempung terpillar TiO₂ mempunyai *basal spacing* 17,80 Å, stabilitas termal pada suhu 200°C, keasaman 2,3575 mmol/gram, keberadaan situs asam Brønsted-Lewis seimbang dan luas permukaannya 169,151 m²/g.

Hasil ini menunjukkan karakter dari lempung terpillar TiO₂ untuk kepentingan adsorpsi atau katalis akan lebih maksimal. Sehingga lempung terpillar TiO₂ siap untuk aplikasi lebih lanjut sesuai kebutuhan yang diinginkan.

SUMMARY

Indonesia has natural materials in the form of clays which have not been exploited optimally. Some research have exploited than becoming better new materials for example as catalyst or adsorbent. By modification of its structure, clays can be processed to be new materials with better physical and chemical properties.

The pillared clay TiO_2 synthesis was conducted by intercalation of titanium solution on clay and followed by calcinations. Calcination temperature was varied out to observe its influence to basal spacing, thermal stability, acidity, Brønsted-Lewis acid sites and surface area of pillared clay.

Characterization of basal spacing and thermal stability used XRD, acidity and Brønsted-Lewis acid sites used pyridine adsorption/IR and surface area used adsorption of nitrogen method through BET equation. Analysis result shown that TiO_2 pillared clay had basal spacing 17,80 Å, thermal stability until 200°C, acidity 2.3575 mmol/g, Brønsted-Lewis acid sites was balanced and surface area 169.151 m²/g.

This result shown character of TiO_2 pillared clay for adsorption or catalyst will be more maximal. So TiO_2 pillared clay is ready to further application according to requirement of needed.

DAFTAR PUSTAKA

- Alberty, R.A.; Daniels, F., 1999, *Pillared Chemistry*, 1st edition, John Wiley, New York, 134 - 140
- Baksh, M.S., Kikkides, E.S. and Yang, R.T., 1992, Characterization by Physisorption of a New Class of Microporous Adsorbents Pillared Clays, *Ind. Eng. Chem. Res.*, 31, 2181 - 2189
- Cheng, L.S., and Yang, R.T., 1995, A New Class of Non Zeolitic Sorbents for Air Separation; Lithium Ion Exchanged Pillared Clays, *Ind. Eng. Chem. Res.*, 34, 2021 - 2028
- Corma, A., 1997, From Microporous to Mesoporous Molecular Sieve Materials and Their Use in Catalysis, *Chem. Rev.*, 2373 - 2419
- Figuera, F., 1988, Pillared Clays as Catalysts, *Catal. Rev. Sci. Eng.*, 30(3), 457 - 499
- Franchi, J.G., Mangialardo, R.C., Lazzari, R.T., Vog, J.C., Fernandez, J.L., Yoshida, R., 1991, In "Industrial Minerals" 92; Ciminelli, R.R., Ed; ABIM; Belo Horizonte, Brazil, 39
- Gillot, J.E., 1987, *Clay in Engineering Geology*, Elsevier, Amsterdam
- Karge, H.G., Hunger, M. and Beyer, H.K., 1999, Characterization of Zeolites Infrared and Nuclear Magnetic Resonance Spectroscopy and X-ray Diffraction, In *Catalysis and Zeolites, Fundamentals and Applications*, Springer, Berlin, 199 - 326
- Kawi, S. and Yao, Y.Z., 1999, Saponite Catalysts with Systematically Varied Mg/Ni Ratio: Synthesis, Characterization and Catalysis, *Micro and Meso Porous Mat*, 33, 49 - 59
- Murtado, H., 1994, Kajian Reaksi Pertukaran Ion Kalsium oleh Ion Natrium pada Bentonit Alam, Skripsi, FMIPA UGM, Yogyakarta
- Ohtsuka, K., 1997, Preparation and Properties of Two-Dimensional Microporous Pillared Interlayered Solids, *Chem. Mater*, 9, 2039 - 2050
- Pinnavaia, T.J., 1983, *Intercalated Clay Catalysts.*, *Science*220, 4595
- Sterte, J., 1986, Synthesis and Properties of Titanium Oxide Cross-linked Montmorillonit, *Clays and Clay Miner.*, 34, 6, 658 - 664

- Takenawa, R., Kemori, Y., and Hayasi, S., 2001, Intercalation of Nitroanilines into Kaolinite and Second Harmonic Generation, *Chem. Mater.*, *13*, 3741 - 3746
- Tan, K.H., 1991, *Dasar-Dasar Kimia Tanah*, a.b. Goenadi, D.H., Gadjah Mada University Press, Yogyakarta, 45 - 165
- Theng, B.K.G., 1974, *The Chemistry of Clay Organic Reaction*, Adam Hogler, London
- Yang, R.T., Chem, J.P., Kikkinides, E.S., and Cheng, L.S., 1992, Pillared Clays as Superior Catalyst for Selective Catalytic Reduction of NO with NH₃, *Ind. Eng. Chem. Res.*, *31*, 1440 - 1445

