

## LAMPIRAN

Tabel Larutan Standar Glukosa Pada Panjang Gelombang 620nm

Sampel	Konsentrasi (ppm)	Absorbansi	Absorptivitas
1	100	0,0016	0,016
2	200	0,003	0,015
3	300	0,0045	0,015
4	400	0,006	0,015
5	500	0,007	0,014

Tabel Larutan Sampel Variabel Berubah F/S Pada Suhu 100°C

Sampel	Suhu (°C)	Absorbansi (620nm)	Konsentrasi Glukosa (mg/l)	Kadar Glukosa (ppm)
1 (1:1)	100	1,740	116	1.242,64
2 (1:2)	100	1,650	110	1.178,35
3 (1:3)	100	1,486	99,06	1.061,21
4 (1:4)	100	1,418	94,53	1.012,64
5 (1:5)	100	1,401	93,4	1.000,5

Tabel Larutan Sampel Variabel Berubah Suhu Pada F/S 1:1

Sampel	Suhu (°C)	Absorbansi (620nm)	Konsentrasi Glukosa (mg/l)	Kadar Glukosa (ppm)
1 (1:1)	60	0,892	59,64	636,92
2 (1:1)	70	1,038	69,2	741,21
3 (1:1)	80	1,130	75,3	806,92
4 (1:1)	90	1,580	105,33	1.128,35
5 (1:1)	100	1,740	116	1.242,64

### ➤ Perhitungan Absorptivitas

$A = a \times b \times c$ . Diketahui tebal kuvet = 1cm

- Sampel 1

$$0,0016 = a \times 1\text{cm} \times 0,1\text{mg/l}$$

$$a = 0,016 \text{ l/mg.cm}$$

- Sampel 2

$$0,003 = a \times 1\text{cm} \times 0,2\text{mg/l}$$

$$a = 0,015 \text{ l/mg.cm}$$

- Sampel 3  
 $0,0045 = a \times 1\text{cm} \times 0,3\text{mg/l}$   
 $a = 0,015 \text{ l/mg.cm}$
- Sampel 4  
 $0,006 = a \times 1\text{cm} \times 0,4\text{mg/l}$   
 $a = 0,015 \text{ l/mg.cm}$
- Sampel 5  
 $0,007 = a \times 1\text{cm} \times 0,5\text{mg/l}$   
 $a = 0,014 \text{ l/mg.cm}$

➤ **Perhitungan Konsentrasi Glukosa pada Larutan Sampel**

$A = a \times b \times c$ . Diketahui tebal kuvet = 1cm, Absorptivitas = 0,015 l/mg.cm

1. Variabel 1

- Perbandingan F/S 1:1 , suhu 100°C  
 $1,740 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$   
 $c = 116 \text{ mg/l}$
- Perbandingan F/S 1:2 , suhu 100°C  
 $1,650 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$   
 $c = 110 \text{ mg/l}$
- Perbandingan F/S 1:3 , suhu 100°C  
 $1,486 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$   
 $c = 99,06 \text{ mg/l}$
- Perbandingan F/S 1:4 , suhu 100°C  
 $1,418 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$   
 $c = 94,53$
- Perbandingan F/S 1:5 , suhu 100°C  
 $1,401 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$   
 $c = 93,4$

2. Variabel 2

- Suhu 60°C, Perbandingan F/S 1:1  
 $0,892 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$   
 $c = 59,64 \text{ mg/l}$

- Suhu 70 °C, Perbandingan F/S 1:1

$$1,038 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$$

$$c = 69,2 \text{ mg/l}$$

- Suhu 80 °C Perbandingan F/S 1:1

$$1,130 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$$

$$c = 75,33 \text{ mg/l}$$

- Suhu 90 °C Perbandingan F/S 1:1

$$1,580 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$$

$$c = 105,33 \text{ mg/l}$$

- Suhu 100 °C Perbandingan F/S 1:1

$$1,740 = 0,015 \text{ l/mg.cm} \times 1\text{cm} \times c$$

$$c = 116 \text{ mg/l}$$

➤ **Perhitungan Kadar Glukosa pada Larutan Sampel**

$$y = ax + b$$

$$y = 0,0014x + 0,0003$$

1. Variabel 1

- Perbandingan F/S 1:1 , suhu 100 °C

$$1,740 = 0,0014x + 0,0003$$

$$x = 1.242,64$$

- Perbandingan F/S 1:2 , suhu 100 °C

$$1,650 = 0,0014x + 0,0003$$

$$x = 1.178,35$$

- Perbandingan F/S 1:3 , suhu 100 °C

$$1,486 = 0,0014x + 0,0003$$

$$x = 1.061,21$$

- Perbandingan F/S 1:4 , suhu 100 °C

$$1,418 = 0,0014x + 0,0003$$

$$x = 1.012,64$$

- Perbandingan F/S 1:5 , suhu 100 °C

$$1,401 = 0,0014x + 0,0003$$

$$x = 1.000,5$$

## 2. Variabel 2

- Suhu 60 °C, Perbandingan F/S 1:1

$$0,892 = 0,0014x + 0,0003$$

$$x = 636,92$$

- Suhu 70 °C, Perbandingan F/S 1:1

$$1,038 = 0,0014x + 0,0003$$

$$x = 741,21$$

- Suhu 80 °C Perbandingan F/S 1:1

$$1,130 = 0,0014x + 0,0003$$

$$x = 806,92$$

- Suhu 90 °C Perbandingan F/S 1:1

$$1,580 = 0,0014x + 0,0003$$

$$x = 1.128,35$$

- Suhu 100 °C Perbandingan F/S 1:1

$$1,740 = 0,0014x + 0,0003$$

$$x = 1.242,64$$

➤ **Perhitungan %Kesalahan**

## 1. Variabel 1

$$y = 0.0091x + 1.026$$

- Sampel 1

$$y = 0.0091 \cdot 116 + 1.026$$

$$y = 2,08$$

$$\%Kesalahan = ((2,08 - 1,74) / 2,08) \times 100 \% = 16,3\%$$

- Sampel 2

$$y = 0.0091 \cdot 110 + 1.026$$

$$y = 2,02$$

$$\%Kesalahan = ((2,02 - 1,65) / 2,02) \times 100 \% = 18,3\%$$

- Sampel 3

$$y = 0.0091 \cdot 99,06 + 1.026$$

$$y = 1,9$$

$$\%Kesalahan = ((1,9 - 1,486) / 1,9) \times 100 \% = 21\%$$

- Sampel 4

$$y = 0.0091 \cdot 94,53 + 1.026$$

$$y = 1,81$$

$$\% \text{Kesalahan} = ((1,81 - 1,418) / 1,81) \times 100 \% = 21\%$$

- Sampel 5

$$y = 0.0091 \cdot 93,4 + 1.026$$

$$y = 1,8$$

$$\% \text{Kesalahan} = ((1,8 - 1,401) / 1,8) \times 100 \% = 22\%$$

$$\begin{aligned} \% \text{ Kesalahan rata-rata} &= (16,3\% + 18,3\% + 21\% + 21\% + 22\%) / 5 \\ &= 19,72\% \end{aligned}$$

## 2. Variabel 2

$$y = 0.0028x + 1.606$$

- Sampel 1

$$y = 0.0028 \cdot 59,64 + 1.606$$

$$y = 1,77$$

$$\% \text{Kesalahan} = ((1,77 - 0,892) / 1,77) \times 100 \% = 49\%$$

- Sampel 2

$$y = 0.0028 \cdot 69,2 + 1.606$$

$$y = 1,79$$

$$\% \text{Kesalahan} = ((1,79 - 1,038) / 1,79) \times 100 \% = 42,5\%$$

- Sampel 3

$$y = 0.0028 \cdot 75,3 + 1.606$$

$$y = 1,8$$

$$\% \text{Kesalahan} = ((1,8 - 1,13) / 1,8) \times 100 \% = 37\%$$

- Sampel 4

$$y = 0.0028 \cdot 105,33 + 1.606$$

$$y = 1,9$$

$$\% \text{Kesalahan} = ((1,9 - 1,58) / 1,9) \times 100 \% = 16\%$$

- Sampel 5

$$y = 0.0028 \cdot 116 + 1.606$$

$$y = 1,93$$

$$\% \text{Kesalahan} = ((1,93 - 1,74) / 1,93) \times 100 \% = 9,8\%$$

$$\begin{aligned} \% \text{ Kesalahan rata-rata} &= (49\% + 42,5\% + 37\% + 16\% + 9,8\%) / 5 \\ &= 30,8\% \end{aligned}$$

## Lampiran Gambar

Keterangan	Daftar Gambar
Larutan Standar Glukosa konsentrasi 100-500 ppm	
Larutan Sampel yang diamati	
Larutan Blanko	
Gambar Spektrofotometer visibel	