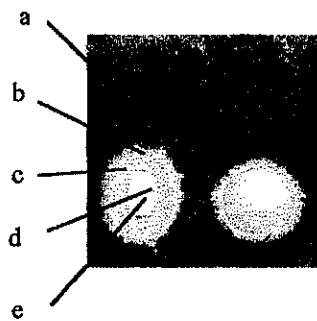
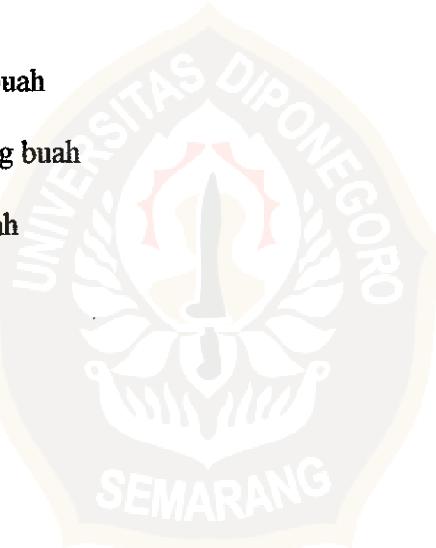


LAMPIRAN A. PENAMPANG MELINTANG BUAH RAMBUTAN

Keterangan:

- a. Kulit luar buah
- b. Kulit daging buah
- c. Daging buah
- d. Kulit biji
- e. Biji buah



Lampiran B. Penentuan Bilangan Asam

$$\text{Bilangan asam} = \frac{b \times M \text{ NaOH} \times 40}{a}$$

a = berat cuplikan (g)

b = mL NaOH yang diperlukan

Penentuan bilangan asam pada lemak yang ditransesterifikasi:

M NaOH = 0,105 M

a = 1 g

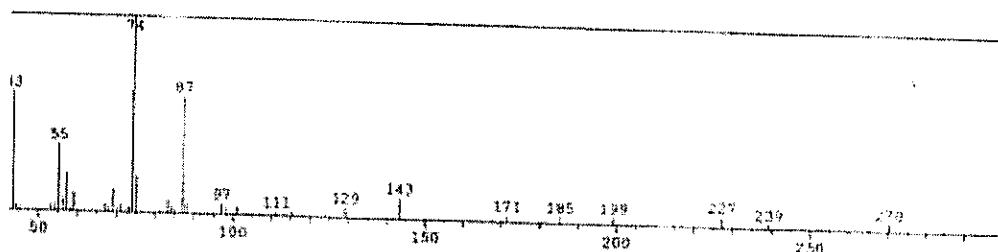
b = 0,9 mL

$$\text{Bilangan asam} = \frac{0,9 \text{ mL} \times 0,105 \text{ M} \times 40 \text{ mg/mmol}}{1 \text{ g}}$$

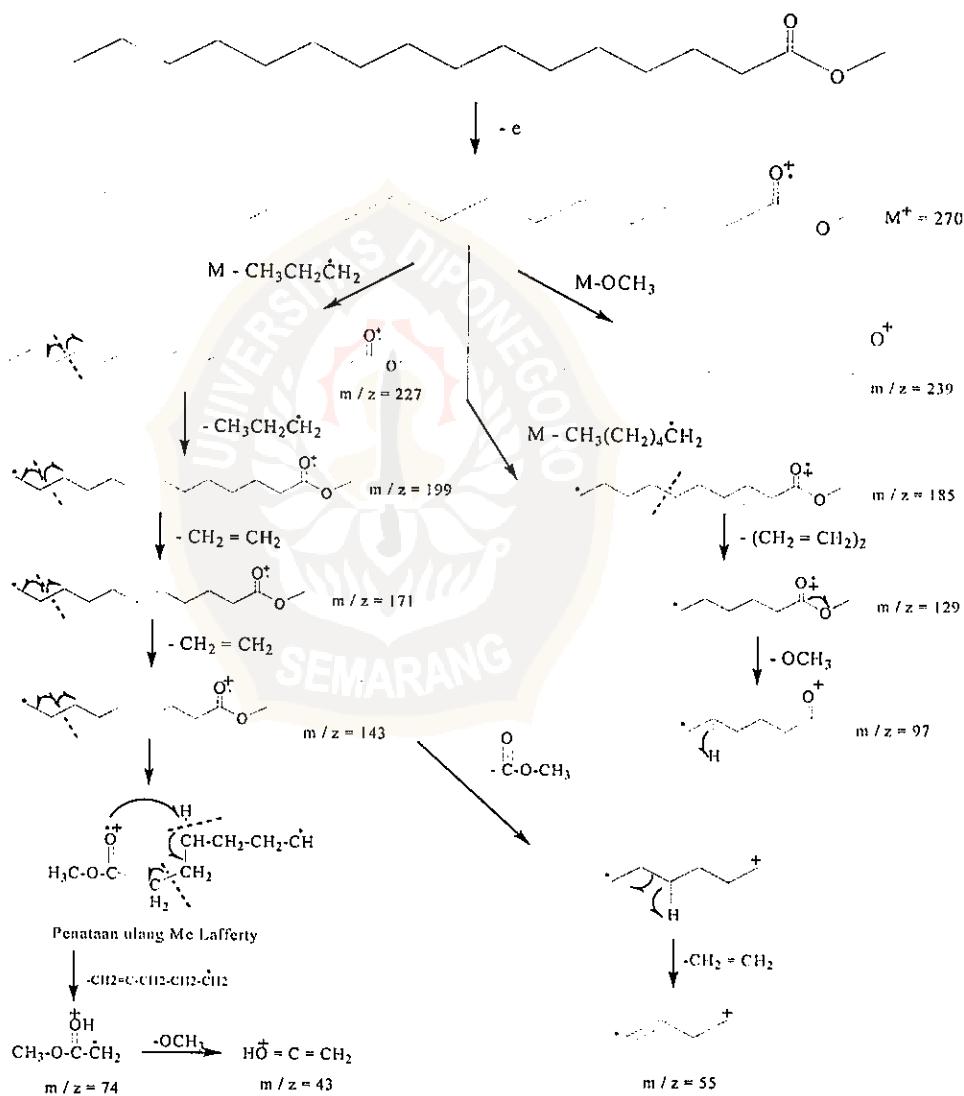
= 3,78 mg NaOH/g lemak

Lampiran C.1. Fragmentasi Spektra Massa Metil Ester Palmitat

Puncak Ke-1

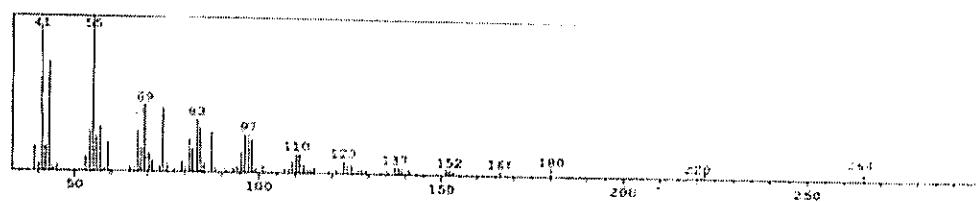


Fragmentasi:

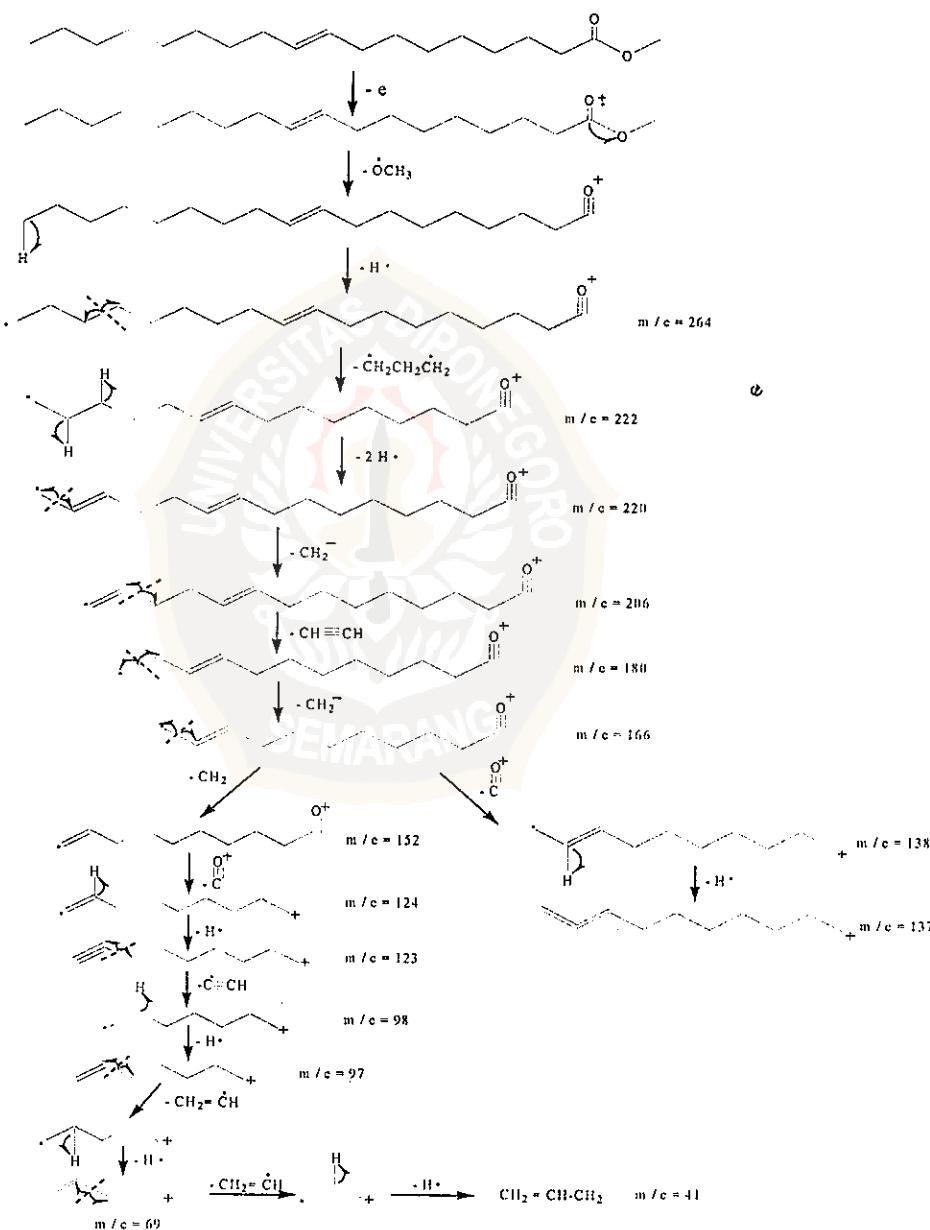


Lampiran C 2. Fragmentasi Spektra Massa Metil Ester Elaidat

Puncak Ke-2

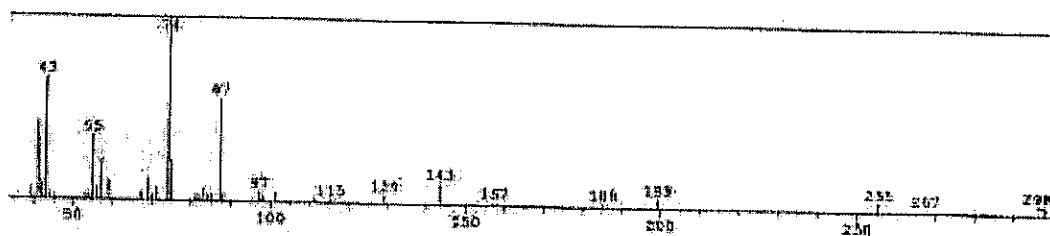


Fragmentasi :

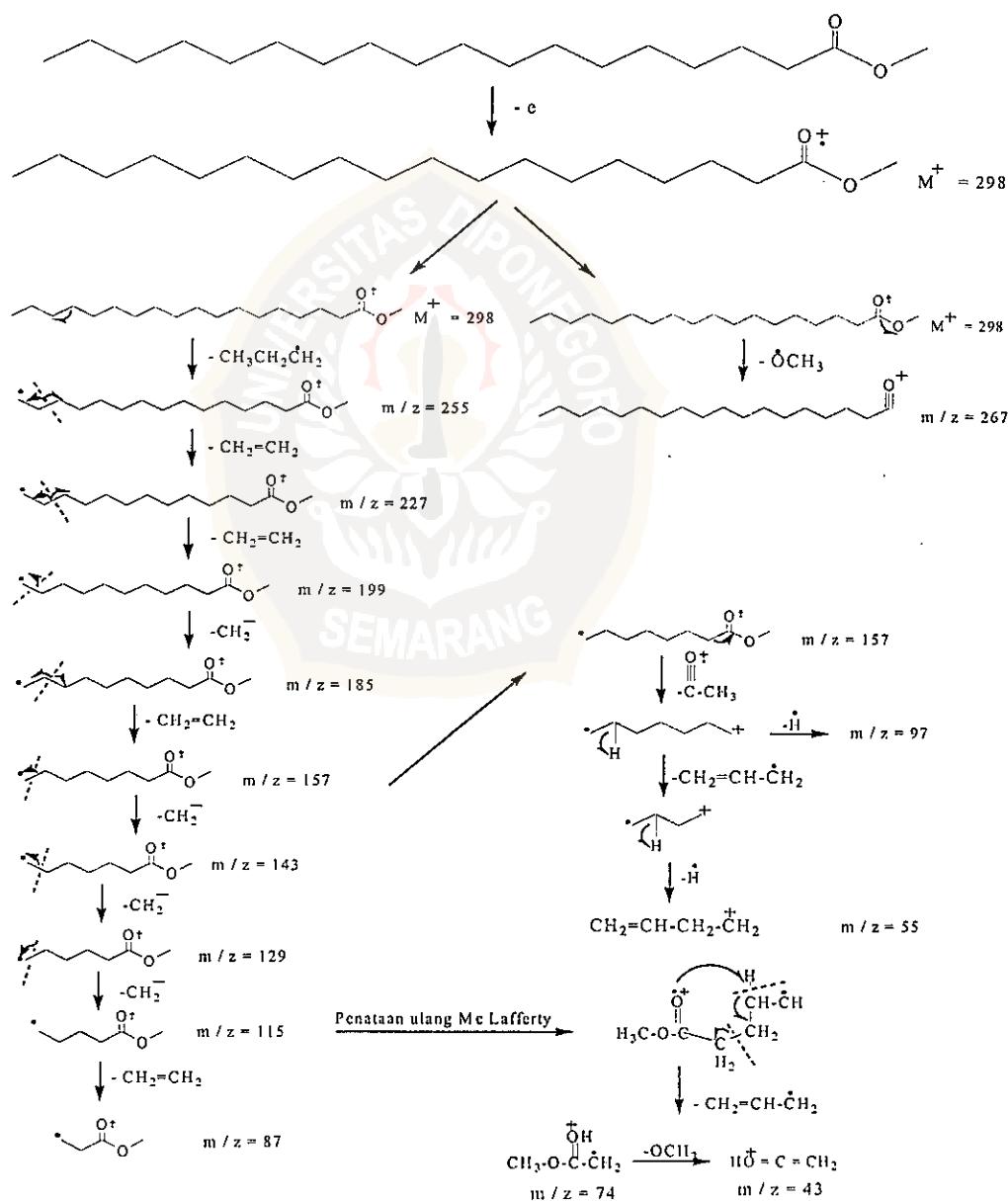


Lampiran C.3. Fragmentasi Spektra Massa Metil Ester Stearat

Puncak ke-3

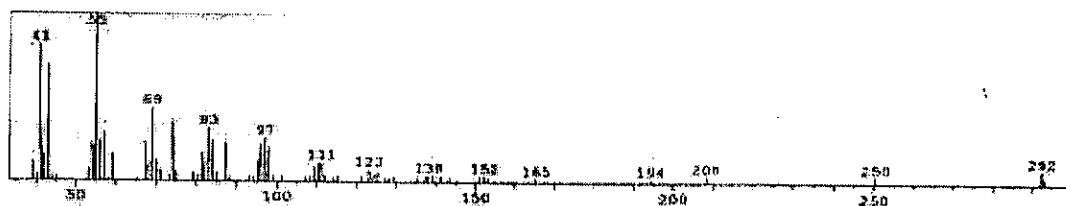


Fragmentasi:

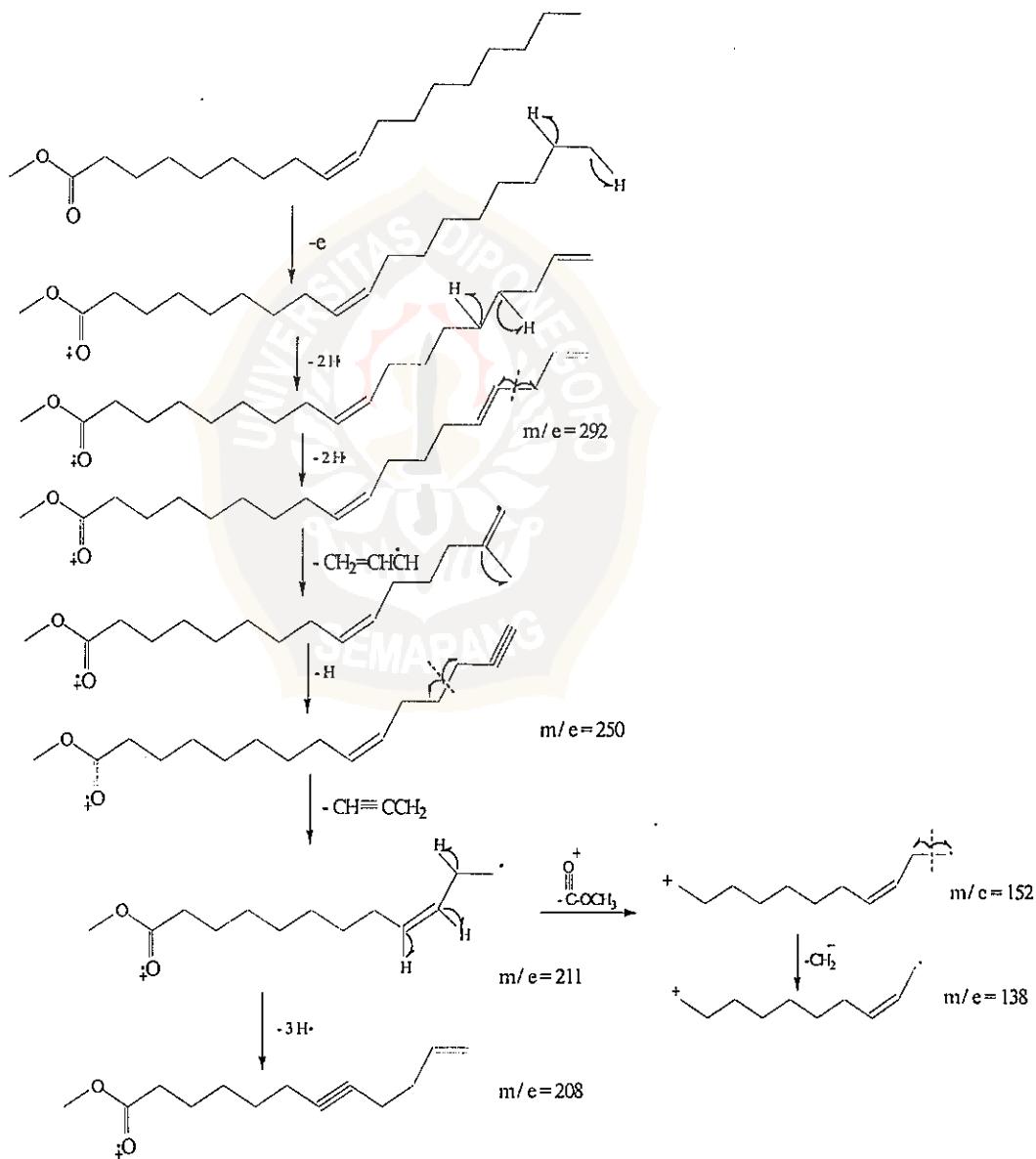


Lampiran C.4. Fragmentasi Spektra Massa Metil Ester Oleat

Puncak ke-4

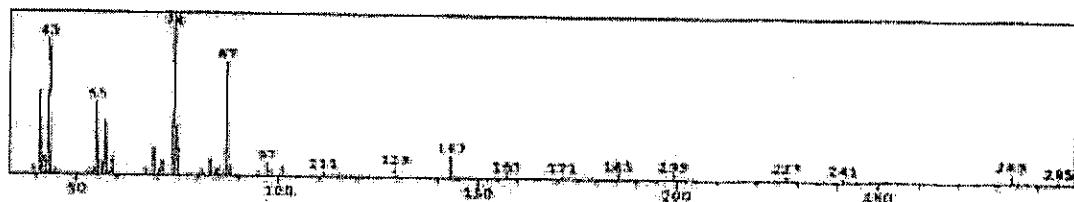


Fragmentasi :

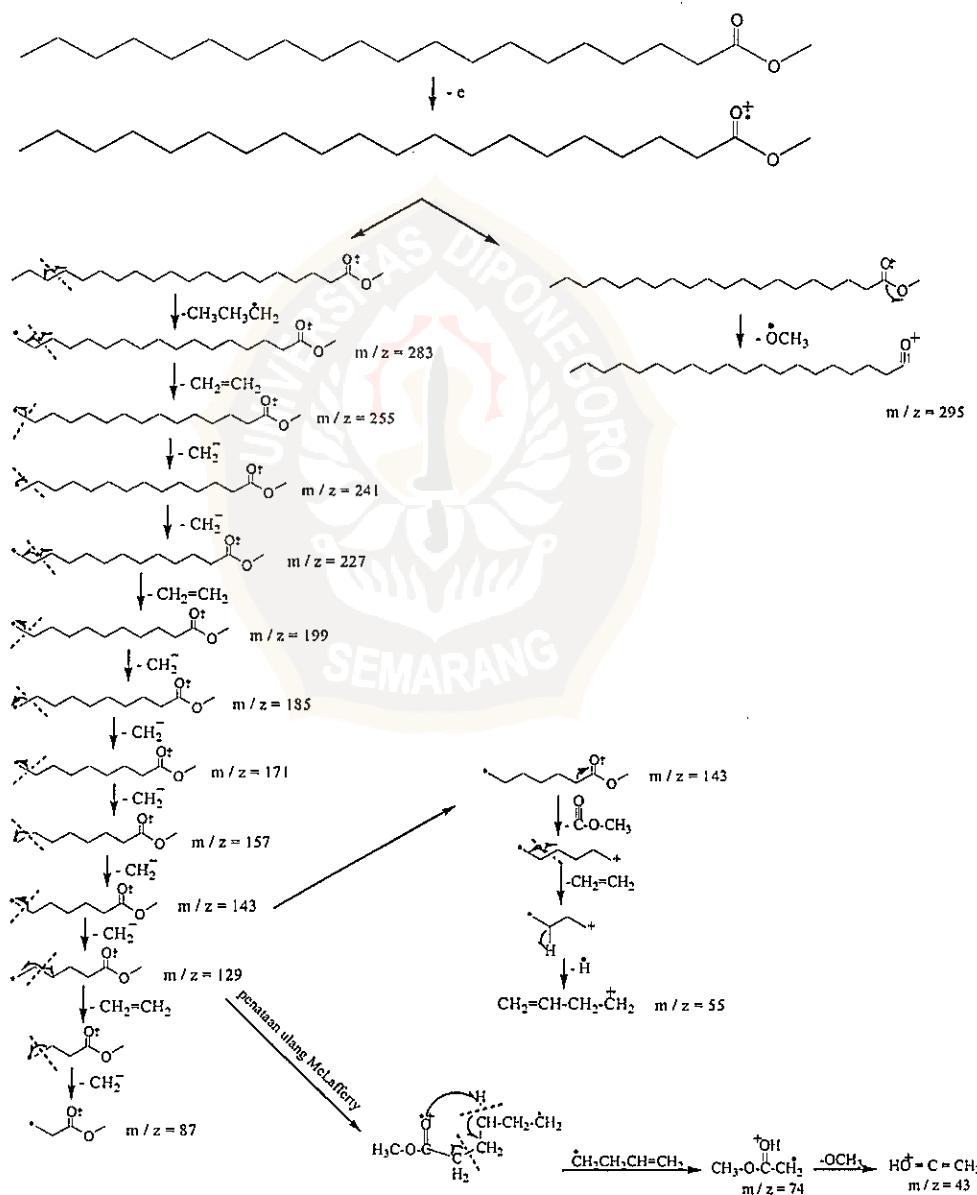


Lampiran C.5. Fragmentasi Spektra Massa Metil Ester Arakhidat

Puncak ke-5

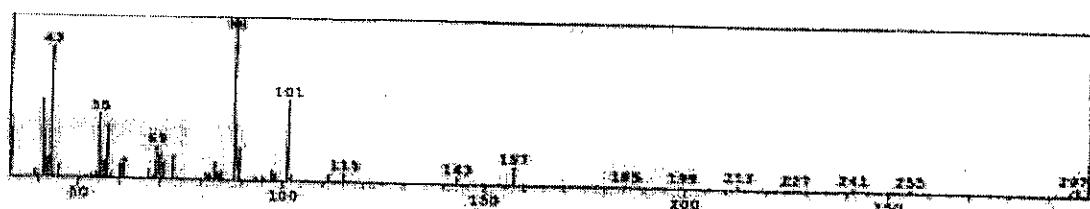


Fragmentasi :

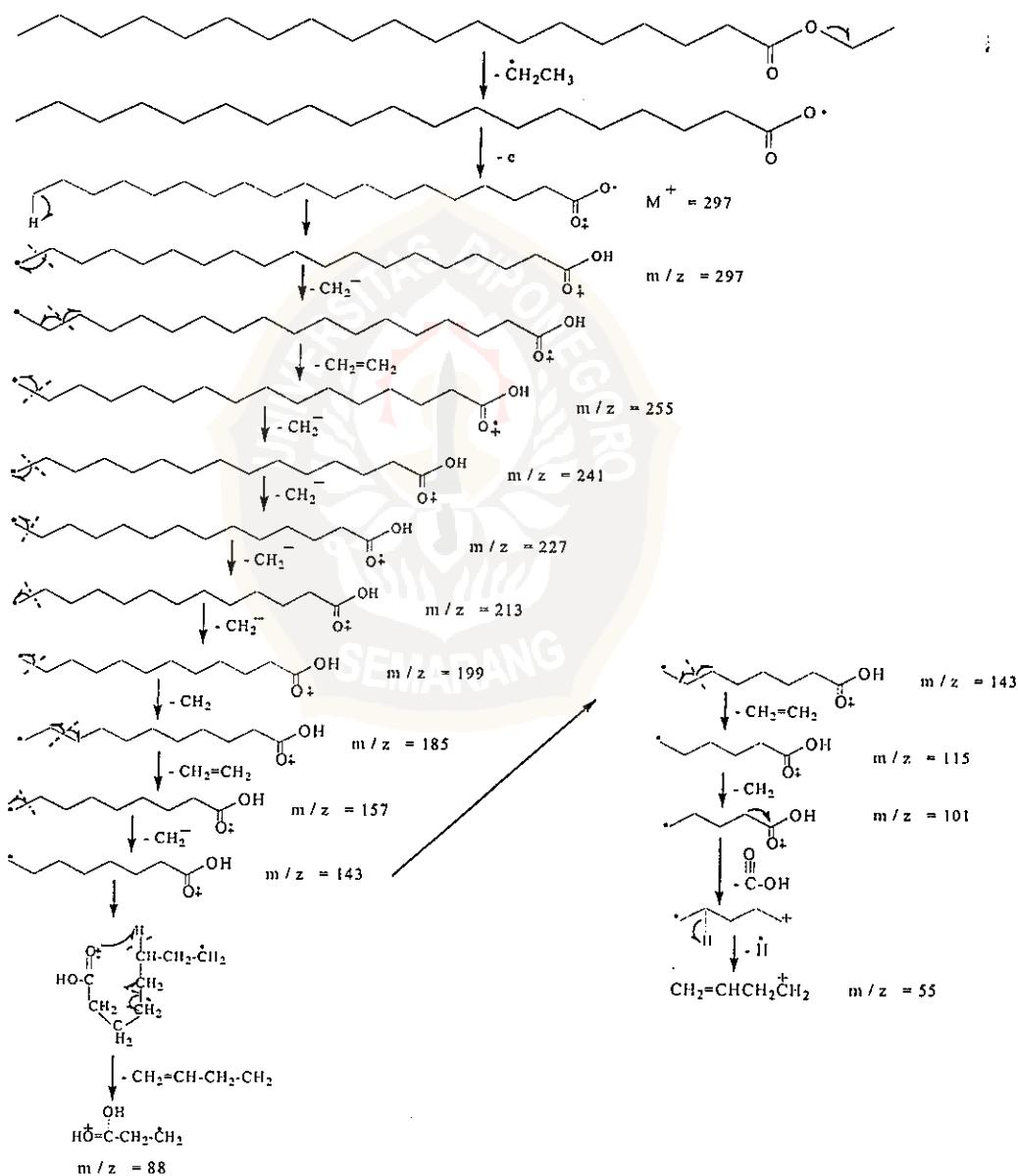


Lampiran C.6. Fragmentasi Spektra Massa Etil Ester Nonadekanoat

Puncak ke-6

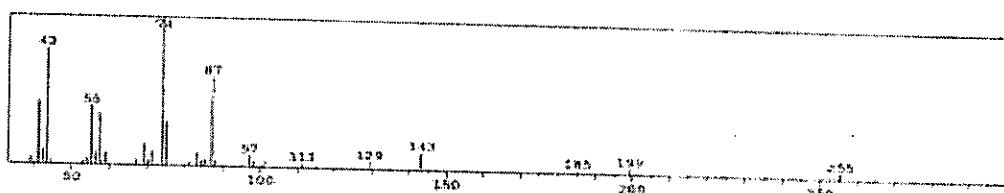


Fragmentasi :

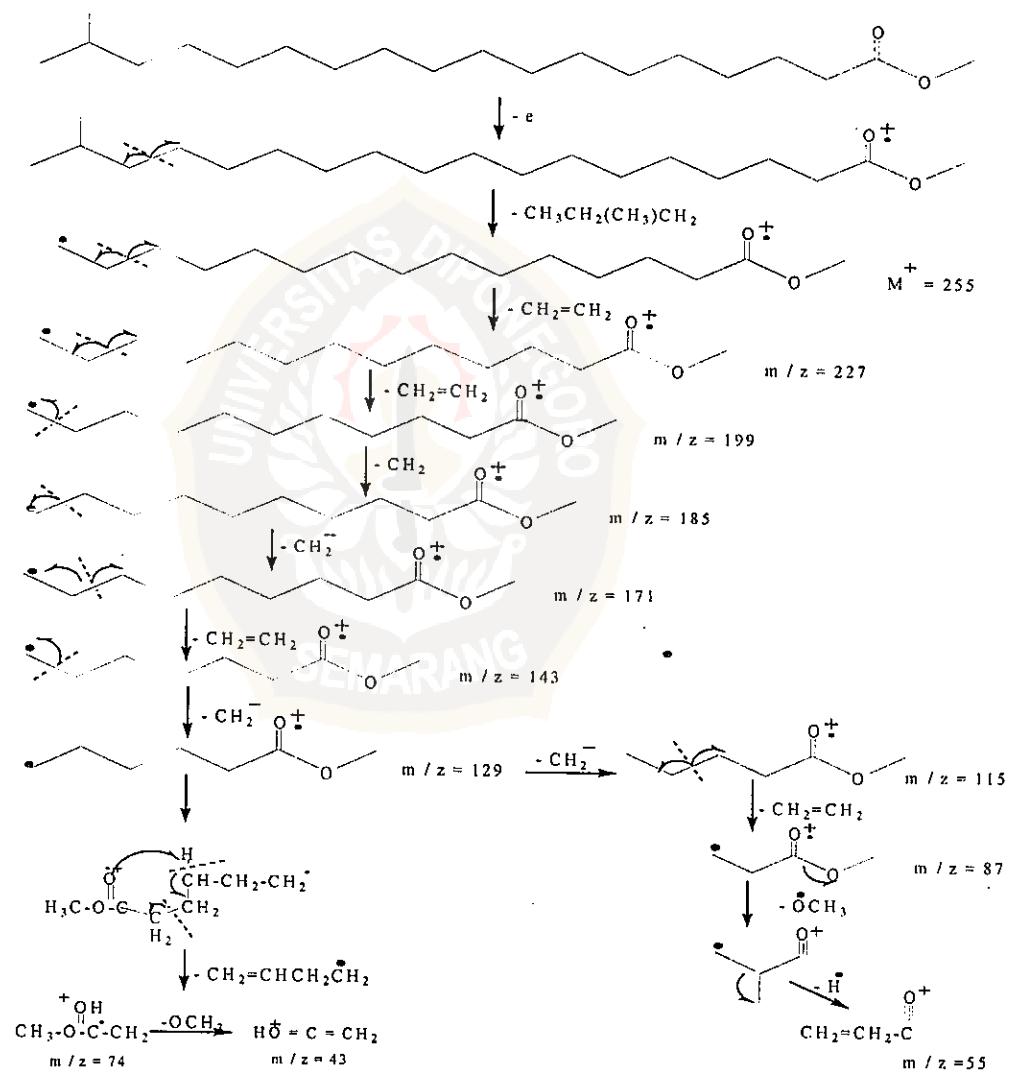


Lampiran C.7. Fragmentasi Spektra Massa Metil Ester 17-Metil Oktadecanoat

Puncak ke-7



Fragmentasi :



Lampiran D. Penentuan Berat Molekul Lemak Biji Rambutan

BM trigliserida = 3 BM asam lemak + BM gliserol – 3 BM H₂O

BM gliserol = 92 g/mol

BM H₂O = 18 g/mol

Asam lemak	BM (g/mol)	Trigliserida	BM (g/mol)	kelimpahan
Asam palmitat	256	Tripalmitin	806	3, 64 %
Asam elaidat	282	Trielaidin	884	29, 78 %
Asam stearat	284	Tristearin	890	5, 33 %
Asam oleat	282	Triolein	884	8, 49 %
Asam arakhidat	312	Triarakhidin	974	44, 69 %
Asam nonadekanoat	298	Trinonadekanoin	932	2, 48 %
Asam 17 metil oktadekanoat	298	Tri-17 metil oktadekanoin	932	5, 58 %

$$\text{BM lemak rata - rata} = \frac{\text{BM lemak} \times \text{kelimpahan asam lemak}}{\text{kelimpahan total}}$$

$$= \frac{806 \times 3,64\% + 884 \times 29,78\% + 890 \times 5,33\% + 884 \times 8,49\% + 974 \times 44,69\% + 932 \times 2,48\% + 932 \times 5,58\%}{3,64\% + 29,78\% + 5,33\% + 8,49\% + 44,69\% + 2,48\% + 5,58\%}$$

$$= \frac{92.548,2}{99,99}$$

$$= 925,57 \text{ g/mol}$$

Lampiran E. Perbandingan Lemak, NaOH, dan Metanol

$$\text{Berat lemak} = 10 \text{ g}$$

$$\text{BM lemak} = 925,57 \text{ g/mol}$$

$$\begin{aligned}\text{Mol lemak} &= \frac{10 \text{ g}}{925,57 \text{ g/mol}} \\ &= 0,0108 \text{ mol}\end{aligned}$$

Perbandingan mol lemak : mol metanol = 1 : 6

$$\begin{aligned}\text{Mol metanol} &= 6 \times 0,0108 \text{ mol} \\ &= 0,0648 \text{ mol}\end{aligned}$$

$$\text{BM metanol} = 32 \text{ g/mol}$$

$$\begin{aligned}\text{Berat metanol} &= 32 \text{ g/mol} \times 0,0648 \text{ mol} \\ &= 2,0744 \text{ g}\end{aligned}$$

Variasi perbandingan persen berat NaOH terhadap lemak : 0,05 %, 0,075 %, 0,1 %, 0,3 %, dan 0,5 %.

Variasi	Berat lemak	Berat metanol	% NaOH (w/w lemak)	Berat NaOH
1.	10 g	2,0744 g	0,05 %	0,005 g
2.	10 g	2,0744 g	0,075 %	0,075 g
3.	10 g	2,0744 g	0,1 %	0,01 g
4.	10 g	2,0744 g	0,3 %	0,03 g
5.	10 g	2,0744 g	0,5 %	0,05 g

Lampiran F. Perhitungan Randemen Metil Ester

Berat lemak yang ditransesterifikasi = 10 g

$$\text{Randemen metil ester} = \frac{\text{berat produk metil ester}}{\text{berat lemak yang ditransesterifikasi}} \times 100\%$$

Variasi NaOH (% w/w lemak)	Berat metil ester	Randemen metil ester
0,05 %	1,1700 g	11,70 %
0,075 %	1,2400 g	12,40 %
0,1 %	1,8507 g	18,51 % *
0,3 %	1,7139 g	17,14 %
0,5 %	1,0380 g	10,38 %

* variasi berat NaOH yang menghasilkan produk optimal