

## LAMPIRAN

### Lampiran 1. Penentuan Kurva Kalibrasi MBAS

Tabel 1.a. Absorbansi Larutan Standart untuk Kalibrasi

Konsentrasi (ppm)	Absorbansi
25	0,388
50	0,398
75	0,420
100	0,440
125	0,470

Tabel 1.b. Data Regresi untuk Kalibrasi

C	A	$(C_i - \bar{C})$	$(A_i - \bar{A})$	$(C_i - \bar{C})(A_i - \bar{A})$	$(C_i - \bar{C})^2$
0	0	-75	-0,425	31,725	5625
25	0,388	-50	-0,035	1,75	2500
50	0,398	-25	-0,025	0,625	625
75	0,420	0	-0,003	0	0
100	0,440	25	0,017	0,425	625
125	0,470	50	0,047	2,35	2500
$\bar{C}=75$	$\bar{A}=0,423$			$\Sigma=36,875$	$\Sigma=11875$

$$\begin{aligned}
 m &= \frac{(C_i - \bar{C})(A_i - \bar{A})}{(C_i - \bar{C})^2} \\
 &= \frac{36,875}{11875} \\
 &= 0,0031
 \end{aligned}$$

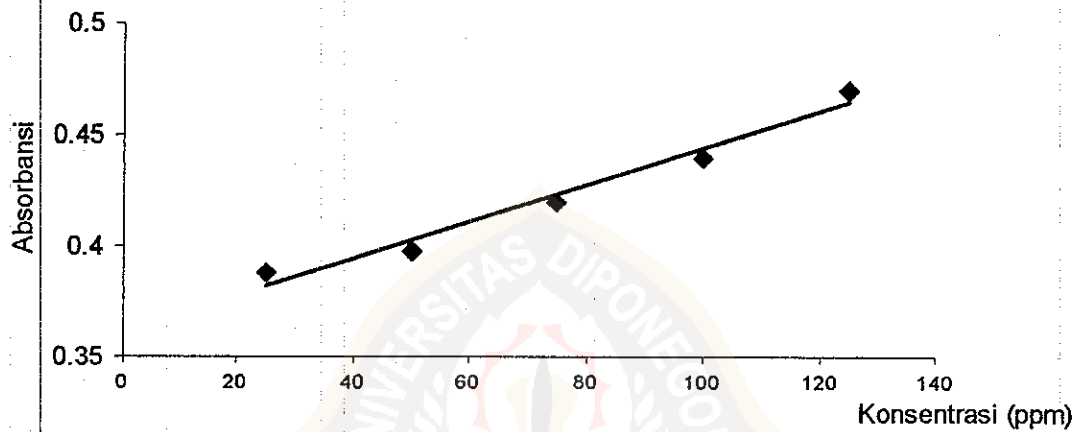
$$A = mC + n$$

$$n = A - mC$$

$$= 0,440 - 0,003 \cdot 100$$

$$= 0,13$$

$$A = 0,0031 C + 0,13$$



Grafik 1. Kurva Kalibrasi Metilen Biru

## Lampiran 2. Perhitungan recovery

### 1. Tanpa Analisis MBAS

- a. Volume tabung pembuat gelembung 11,19 cm<sup>3</sup>

$$m = 0,201 \text{ gr}$$

$$\text{Sehingga konsentrasinya} = \frac{201 \text{ mg}}{1 \text{ L}} = 201 \text{ ppm}$$

$$\text{Recovery} = \frac{201}{1000} \times 100 \% = 20,1 \%$$

- b. Volume tabung pembuat gelembung 21,98 cm<sup>3</sup>

$$m = 0,290 \text{ gr}$$

$$\text{Sehingga konsentrasinya} = \frac{290 \text{ mg}}{1 \text{ L}} = 290 \text{ ppm}$$

$$\text{Recovery} = \frac{290}{1000} \times 100 \% = 29,0 \%$$

### 2. Dengan Analisis MBAS

- a. Volume tabung pembuat gelembung 11,19 cm<sup>3</sup>

$$[\text{surfaktan}] = 75,80 \text{ ppm}$$

$$\text{Recovery} = \frac{75,80}{1000} \times 100 \% = 7,58 \%$$

- b. Volume tabung pembuat gelembung 21,98 cm<sup>3</sup>

$$[\text{surfaktan}] = 100,65 \text{ ppm}$$

$$\text{Recovery} = \frac{100,65}{1000} \times 100 \% = 10,06 \%$$

### Lampiran 3. Perhitungan Tegangan Permukaan

$$\text{Tegangan Permukaan } (\gamma) = \frac{r \cdot g \cdot h \cdot \rho}{2}$$

Diketahui :

$$h_{\text{air}} = 3,2 \text{ cm}$$

$$h_{11,19} = 2,7 \text{ cm}$$

$$h_{21,98} = 2,4 \text{ cm}$$

$$\gamma_{\text{air}} = 72 \text{ dyne/cm}$$

$$\rho_{\text{air}} = 1 \text{ g/mL}$$

$$r = \frac{2 \cdot \gamma}{g \cdot h \cdot d}$$

$$r = \frac{2 \cdot 72 \text{ dyne/cm}}{980 \text{ cm/s}^2 \cdot 3,2 \text{ cm} \cdot 1 \text{ g/mL}}$$

$$r = 0,0459 \text{ cm}$$

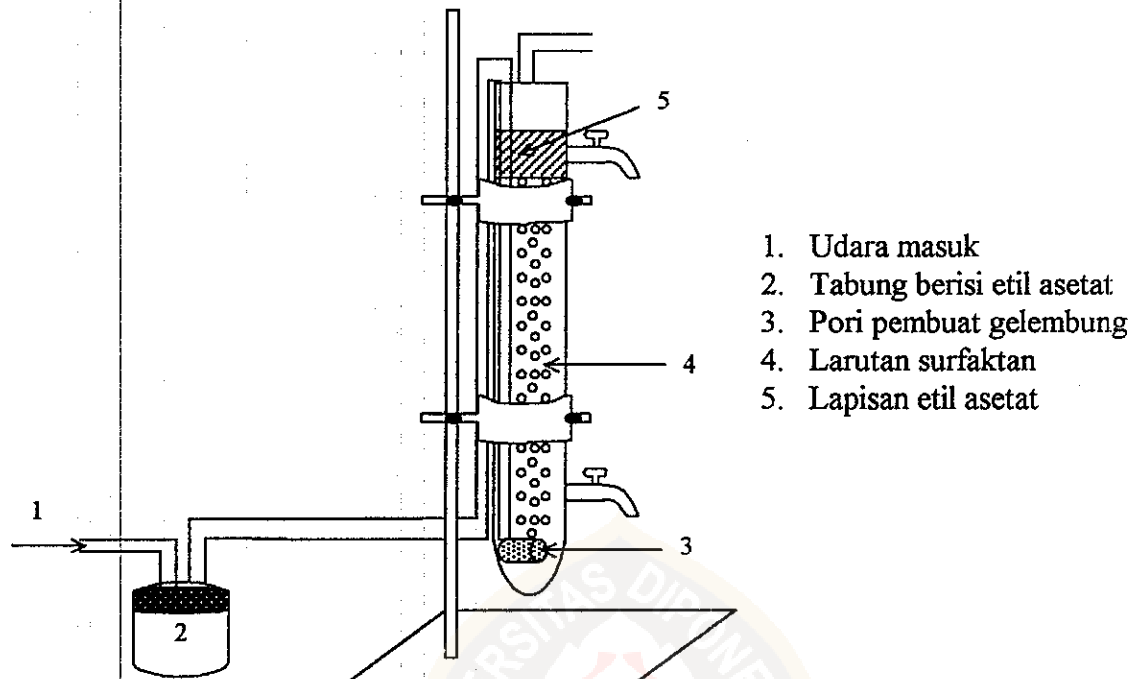
$$\begin{aligned} \gamma_{11,19} &= \frac{r \cdot g \cdot h \cdot \rho}{2} \\ &= \frac{0,0459 \cdot 980 \cdot 2,7 \cdot 1}{2} \end{aligned}$$

$$= 60,8 \text{ dyne/cm}$$

$$\begin{aligned} \gamma_{21,98} &= \frac{r \cdot g \cdot h \cdot \rho}{2} \\ &= \frac{0,0459 \cdot 980 \cdot 2,4 \cdot 1}{2} \end{aligned}$$

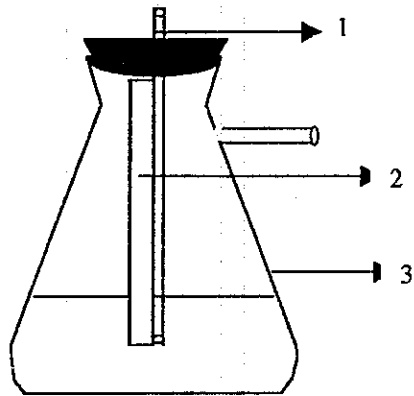
$$= 54,1 \text{ dyne/cm}$$

## Lampiran 5. Desain Alat Proses Sublasi



Gambar 1. Desain Alat Proses Sublasi

## Lampiran 6. Alat Pengukur Tegangan Permukaan



Keterangan :

1. Pipa Kapiler
2. Penggaris
3. Erlenmeyer

Gambar 2. Alat Pengukur Tegangan Permukaan

